

Acid-Stimulated Biological Phosphorus Removal

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Summary

- QUESTOR Centre
- Microbial phosphorus (P) cycling
- Exploit microorganisms to remove and recover P
- Polyphosphate metabolism

QUESTOR Centre

Water & Wastewater Cluster

- Cluster covers all aspects of water / wastewater treatment
- All treatment processes (biological, chemical, physical, thermal etc.)
- Pollution prevention, reuse/recycling
- Effects of discharges on the environment
- Reduction/elimination of associated sludges/residuals
- Public Health Issues
- Sustainability – energy, carbon emissions
- Pathogen tracking

Membership

- McAllister Brothers Ltd.
- ExxonMobil R. & E.
- Enva Ireland Limited
- Central Chemical Supplies Ltd.
- T.E. Laboratories Ltd.
- Modern Water plc
- Wilson's Country Ltd.
- Northern Ireland Water
- Glenfarm Holdings Ltd.
- Williams Industrial Services
- Cleanfields Technologies Ltd.
- Colloide Engineering Systems
- Bombardier Aerospace
- Chevron North Sea Ltd.
- Northern Innovation Ltd.
- NI Environment Agency

Membership

- Queen's University Belfast
- IWW Water Centre
- University of Duisburg Essen
- Dalhousie University
- Stevens Institute of Technology
- Cranfield University
- Dublin City University

Phosphorus (P)

- P one of the most essential nutrients present within the biosphere
- Found in the environment in both inorganic form (as orthophosphate; P_i) and in a variety of organic compounds
 - organophosphates (which contain a C-O-P ester bond)
 - organophosphonates (which contain a direct C-P bond and are often potent antimetabolites)

P an Introduction



There was taken a considerable quantity of Man's urine (because the liquor yields but a small proportion of the desired Quintessence) and of this good part at least, had been for a pretty while digested before it was used. Then this Liquor was distilled with a moderate Heat, till the Spiritous and Saline parts were drawn off

Introduction



The Alchemist in Search of the Philosopher's Stone (1771) by Joseph Wright



Introduction



- P is a scarce natural resource for which there is intense competition between life forms
- Those organisms which can scavenge, store and recycle P have a natural advantage
- Unlike nitrogen which can be replenished from atmospheric sources, the only natural source of P is from rock
- Conserving and re-using this resource is important

Discovery of phosphorus

Robert Boyle or Hennig Brandt

Albright

Sir John Laws developed superphosphate and founded the oldest agricultural research station in the world – Rothamsted, where some of his classical experiments are still being maintained.

Adults excrete 98% of the P in their diets (because they are turning over cells) this ends up in urban wastewater. It can be responsible for eutrophying inland waters.



Eutrophication



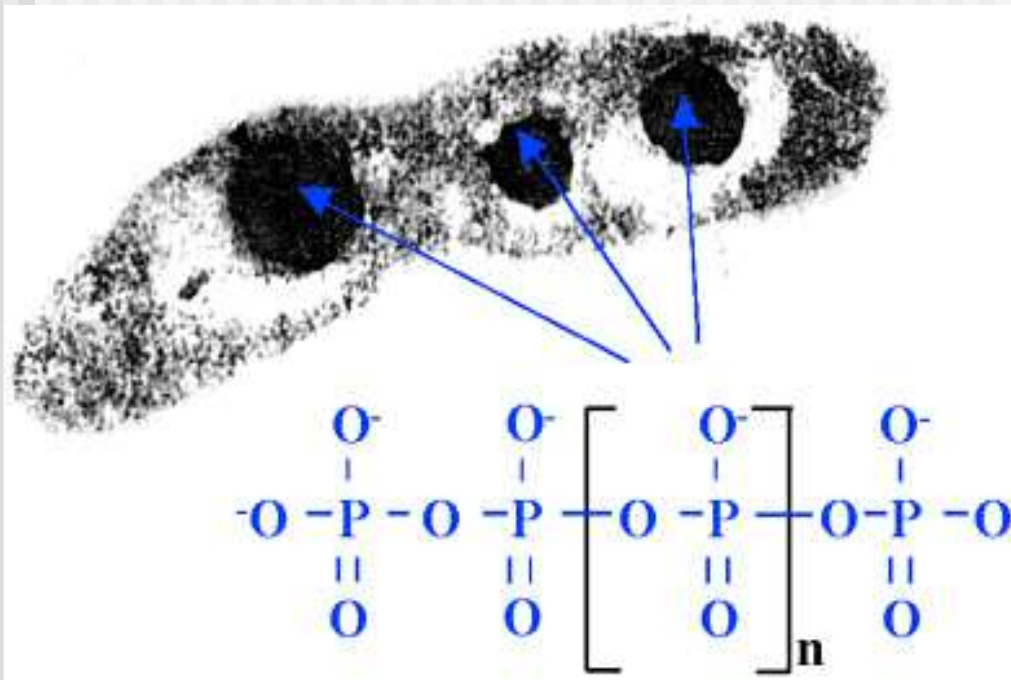
Compliance Limits

- Urban Wastewater Treatment Directive
- Designates areas as “eutrophication sensitive”
- Requires a sewage treatment works to remove 80% of the phosphorus coming in
- Normally a works will only remove 40%

Chemical Precipitation: Disadvantages

- Cost of precipitants
- Need for tertiary filtration due to potential for heavy metal contamination
- Necessity to dispose of excess volumes of sludge (sludge volumes may be increased 20 - 40% through chemical P precipitation)

Polyphosphate



- Consists of a linear chain of phosphate residues ranging in length from 3 to greater than 1000 residues
- Molecular fossil
- Wide range of biological roles

The reason why cells will accumulate polyphosphate is unknown. Some pathogens lose their virulence if their ability to accumulate poly-P is removed.

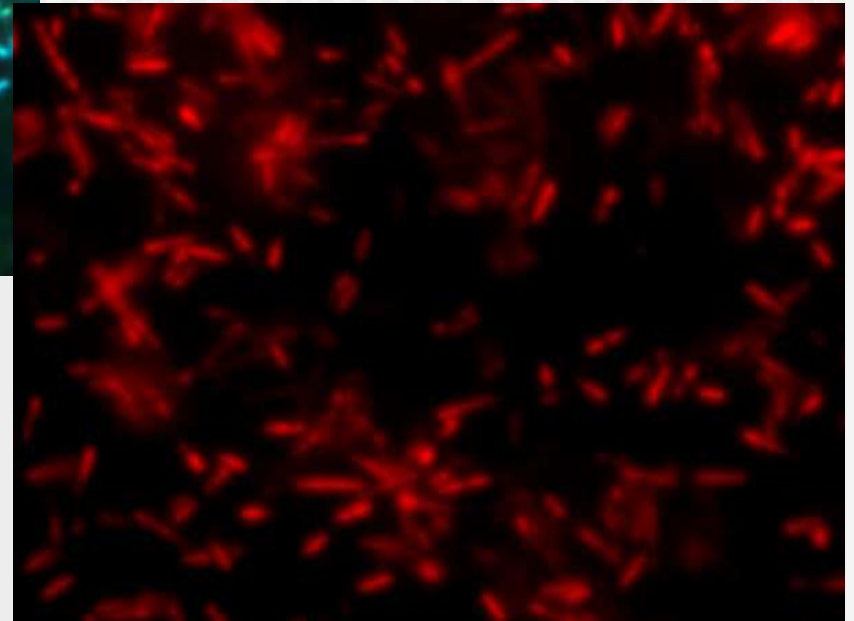
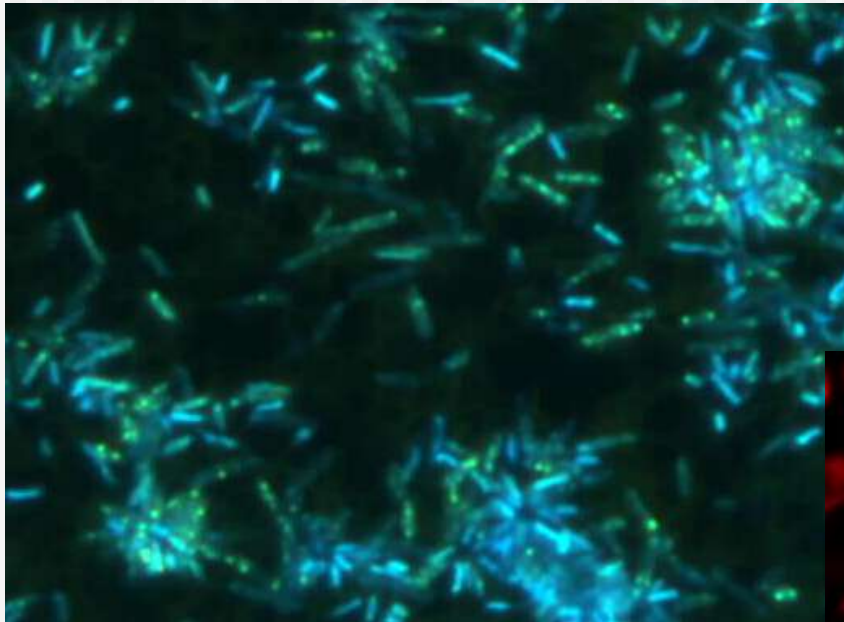
Research: Polyphosphate metabolism

- Marine environment – involved in survival under varying nutrient conditions
- Medical – pathogenicity and virulence
- Stain any microbial sample visualise polyP
- Reservoir of energy and phosphate, chelator of metals, capsule material, and in the adaptation of microbial cells to growth and development
- Role in the response of microbial cells to nutritional and environmental stresses.

Enhanced Biological Phosphorus Removal

- Principle of EBPR is the cycling of the microbial biomass through anaerobic and aerobic stages
- Under favourable conditions EBPR plants can remove 80 - 90 % of influent phosphate
- Conventional activated sludge treatment removes only 20 - 40%

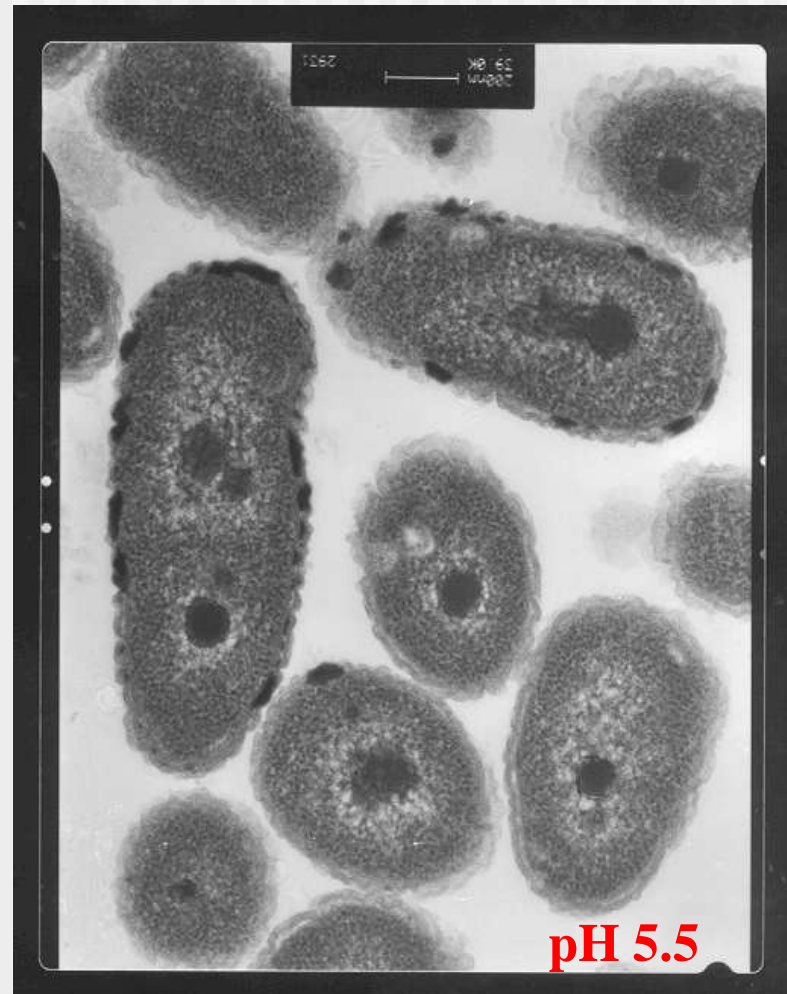
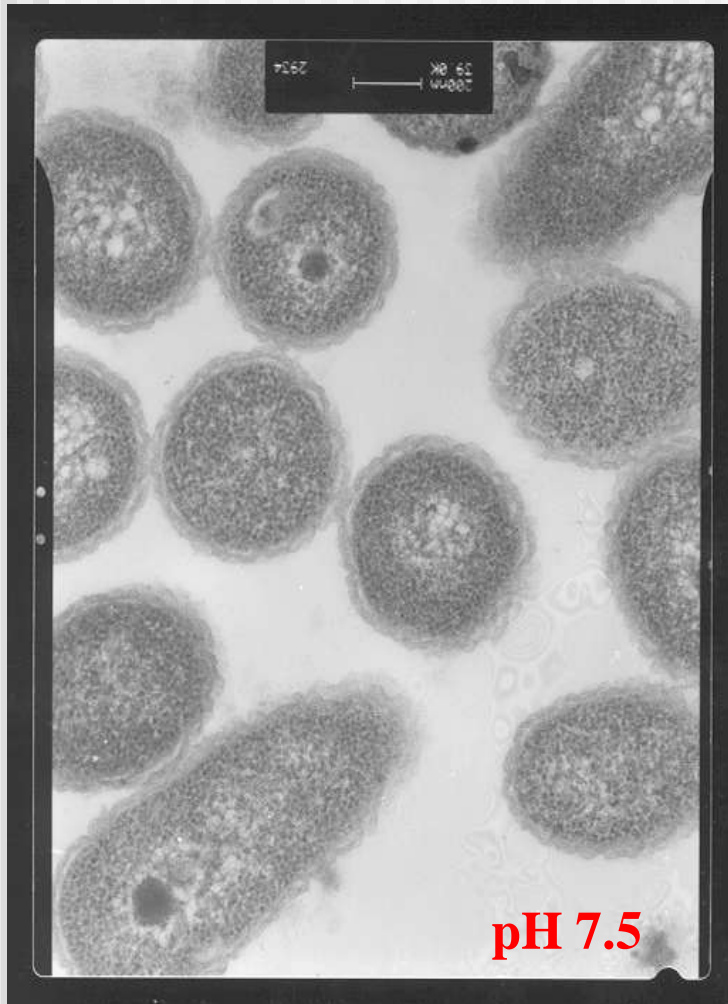
Enhanced Biological Phosphorus Removal



Rationale

- Anaerobic/aerobic cycling of EBPR elicits polyP response
- Accumulation of polyP be triggered by other stresses
- Identify a variety of conditions under which the ability to accumulate polyP is necessary for microbial survival
- Accumulation of polyP, by its very nature, will result in enhanced phosphate removal.
- Such conditions could be exploited to provide alternative treatment options for biological P removal.

Burkholderia cepacia AM19



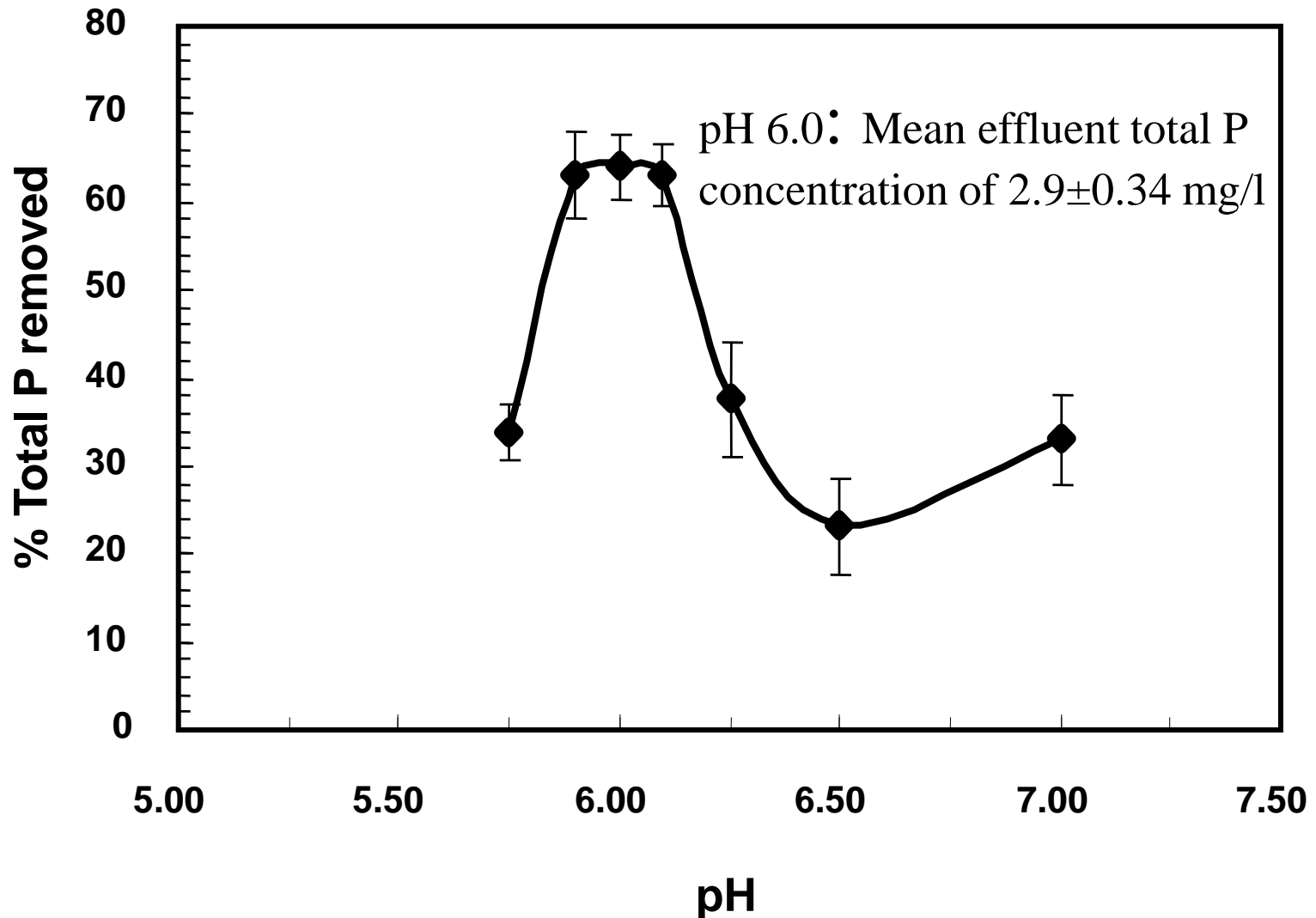
2000 litre pilot plant at New Holland WwTW, Belfast. 2 l/min inflow. Worked well. pH was controlled. HRT 15 h; sludge age 10-14 days.

Pilot Plant

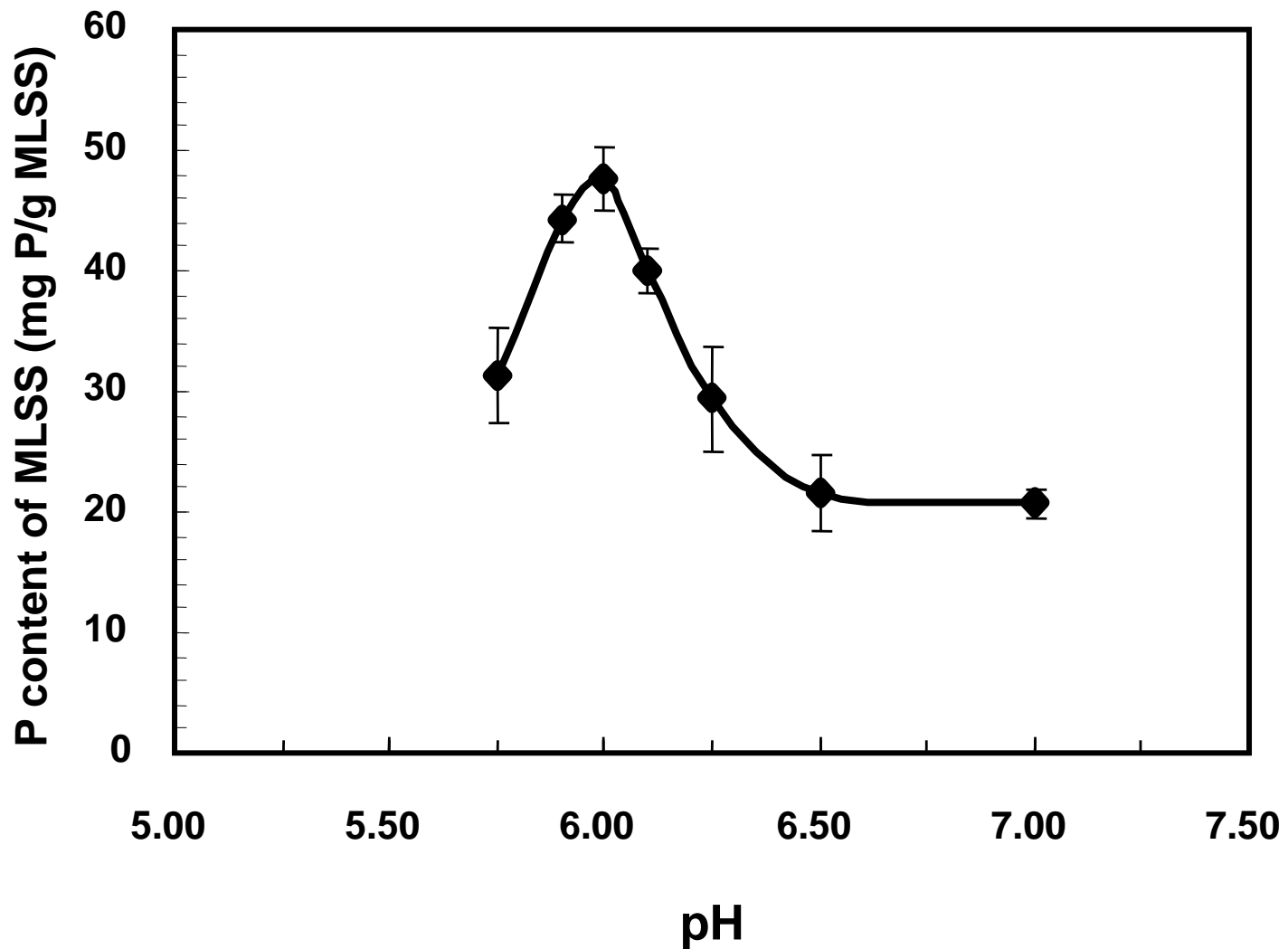




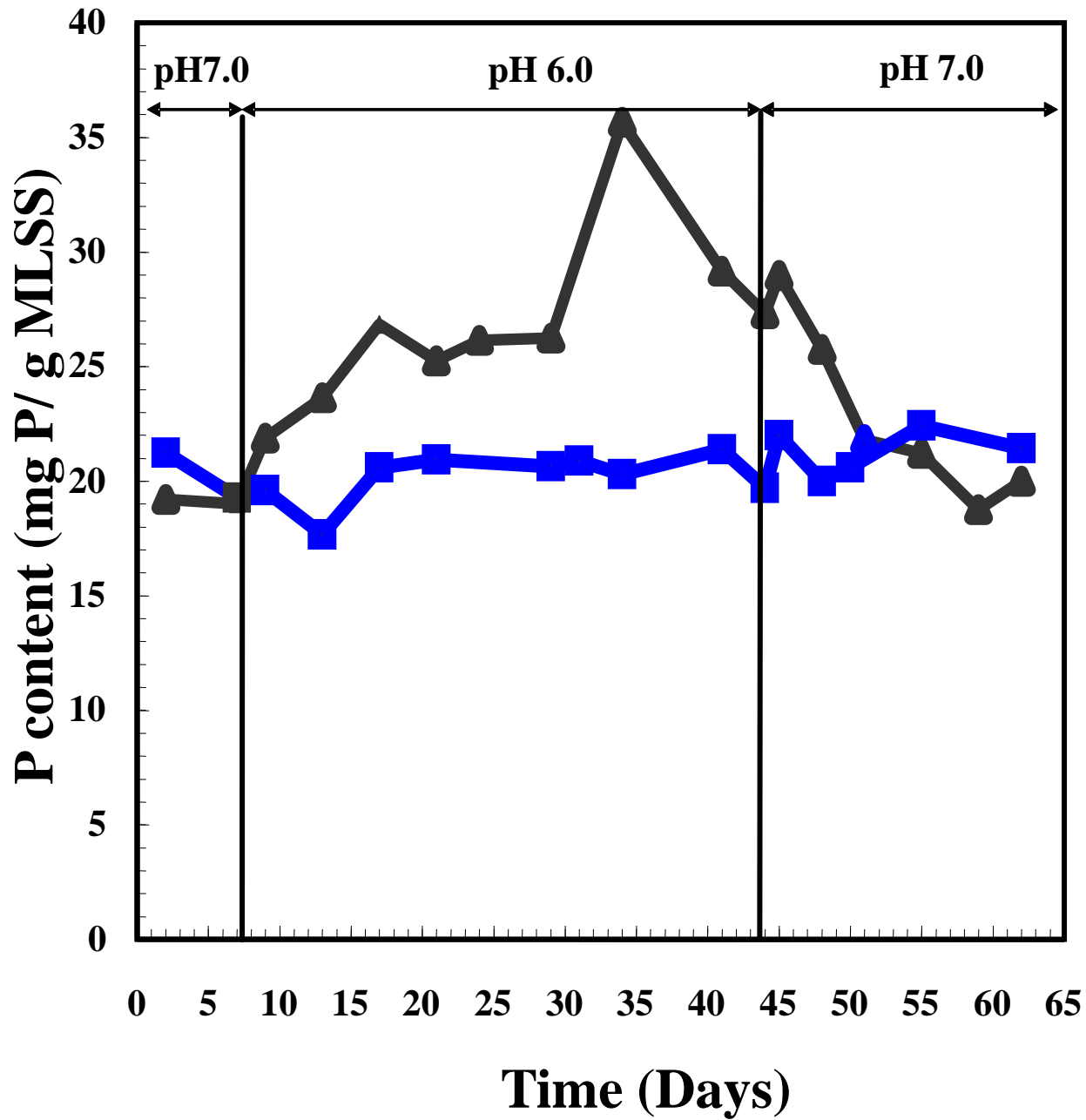
- Influent characteristics: Total P 6-14mg/L, $\text{NH}_3\text{-N}$ 15-25mg/L COD 200-500mg/L.



- Main New Holland Works pH 7.0 - 7.3: $34 \pm 5\%$ (effluent total P concentration 5.9 ± 0.3 mg/l)



- Main New Holland Works pH 7.0 - 7.3: 20±3 mg P/g MLSS



Pilot
Plant

Main
Works

Pilot Plant Summary

- Operation of pilot plant at pH 6.0 achieve a total effluent P concentration of 2.9mg/l
- Equates to 64% total P removal
- UWWT Directive requires an effluent total P standard of:
 - 2.0 mg/l for STW serving up to 100,000 p.e.
 - 1 mg/l for installations greater than 100,000 p.e.
 - 80% P removal across the system as a whole.

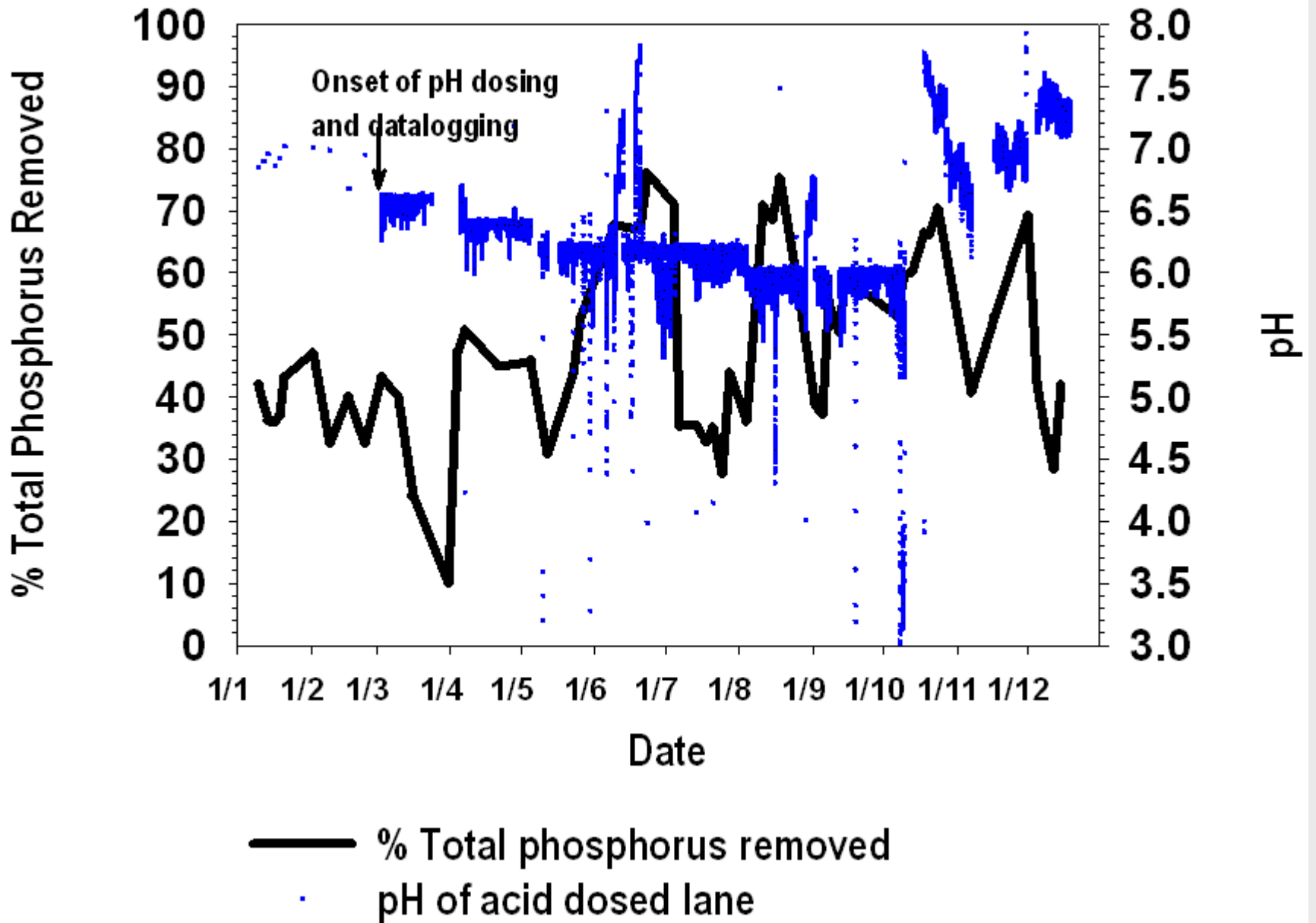
Pilot Plant Summary

- Not yet in a position to claim compliance
- Limitations of the pilot plant system e.g. continuous influent flow rates
- **Inconsistent**
- Also only if there is a firm understanding of its scientific basis
- pH 6.0 total P removal increases by approximately 31%

Mulduglass is a very small WwTW, with very frequent rain.

Full Scale Trial





Full Scale Trial



Flushed bra causes sewer collapse



A bra and a pair of knickers have been blamed for a flood and road collapse in County Durham. Northumbrian Water said the underwear was flushed down a toilet and caused a blockage in a sewage pipe in Middleton-St-George, near Darlington.

- A spokeswoman said: "If the underwear had not been flushed down the toilet, this would not have happened. It was very irresponsible behaviour".
- "These pipes are not designed to carry bras and knickers." (BBC, 19 June 2007)

Implications for P removal

- In contrast to EBPR P accumulation occurs in a fully aerobic environment
- Uptake does not appear dependent on presence of VFAs
- Existence of a widespread natural phenomenon, whereby exposure to stress elicits a polyP accumulation response.

35% of the bugs in activated sludge exhibit poly-P accumulation if under pH stress. Nitrification performance dips at pH 6.8 but then recovers as pH decreases further. It is conceivable that a granulated bacterial with acid producers on the outside (nitrifiers) and poly-P accumulators on the inside (or layered biofilms) in the manner of ANAMMOX granular biomass could be effective.

Phosphorus: the disappearing nutrient

- 161 million tonnes of P are extracted each year
- Current reserves to last only next 45 - 100 years
- P supply concentrated in China, Morocco, the US and Russia: more than 70% of the global P deposits
- P prices – 2008 \$500 per tonne, five fold higher than 2007
- Oil: alternatives can be found. No substitute for P.
- Biggest P gains from recovery - recycling of P from waste
- Canadian study: P recovery from activated sludge could meet 30% of its current needs
- For this to be viable high sewage P content.

Current Research

- Research to develop in a number major areas:
- Recently identified additional stress which significantly increases P removal
- Other stresses which may elicit the polyP response



Current Research

- P recovery: Can we release this stored P from the cells?
 - ▶ In what form is this P stored?
 - ▶ Is it suitable for recovery?
 - ▶ What triggers its degradation / release from cells?
 - ▶ Can we capture the released P or indeed polyphosphate?
 - ▶ Can it be scaled – using 2 litre and 20 litre bioreactors?



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